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Title: Base station lead-acid battery discharge curve

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What happens when a lead acid battery is discharged?

In lead-acid cells, the electrolyte (sulfuric acid) participates in the cell's normal charge/discharge reactions. As the cells are discharged, the sulfate ions are bonded to the plates-- sulfuric acid leaves the electrolyte.

What is a lead-acid battery?

Lead-acid batteries are widely consumed in the automotive industry, as a source of energy in automotive vehicles, and also in large-scale systems such as electric power supply. For these main reasons, the lead-acid battery is the type of battery to be studied.

What is the SOC profile of a lead-acid battery?

The discharge voltage curves may be depressed by as much as 0.5 VDC from those shown on the graph. Charge voltages will be elevated by lead-acid battery. In lead-acid cells, the electrolyte (sulfuric acid) participates in the cell's normal charge/discharge reactions.

Which variables are constant in a lead acid battery?

Battery temperature, average DOD, and battery lifetime are all of these variables that have been considered to be constant for the obtained experimental data here. A mathematical model has been formulated and verified with experimental data to describe a lead acid battery's discharging and charging characteristics here.

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